Photoisomerization of 2-Cyanopyrrole

By H. HIRAOKA

(IBM T. J. Watson Research Center, Yorktown Heights, New York 10598)

Summary Irradiation of 2-cyanopyrrole in methanol yields 3-cyanopyrrole as a major photo-product; this is believed to be the first example of the 2-3 valance isomerization of pyrroles.

PHOTOISOMERIZATIONS involving transfer of a substituent from the 2-position to the 3-position have been reported in the isoxazole,¹ furan,² thiophen,³ and pyrazole⁴ series. The specific 2-3 direction of photoisomerization has been correlated with the bond orders of the excited ring systems calculated from simple Hückel-type molecular orbitals.⁵ In spite of these findings, the 2-3 valence isomerization of pyrroles has not previously been reported, though thermal⁶ and photochemical⁷ 1-2 and 1-3 valence isomerizations have been reported. In this last case, however, the 2-isomer is not the precursor of the 3-isomer, because 2-benzylpyrrole does not isomerize to its 3-isomer either photochemically or thermally.

I report that 2-cyanopyrrole photoisomerizes to 3-cyanopyrrole in a methanol solution on irradiation with u.v. light, and that 3-cyanopyrrole does not photoisomerize to the 2-isomer under the same conditions.

2-Cyanopyrrole was first of all prepared by dehydration

of pyrrole-2-aldoxime by acetic anhydride,⁸ but it was found that its photo-dehydration in methanol was more convenient.[†] The i.r., n.m.r., and molecular weight of the 2-cyanopyrrole thus obtained agreed completely with those in the literature.9

A 0.02% solution of 2-cyanopyrrole in methanol in a quartz tube, after thorough degassing, was placed in the centre of a bank of low-pressure mercury lamps, and was irradiated for 20 hr. The solvent was then removed by distillation. The photo-products were isolated by g.l.c. with a 6 ft Apiezon oil column at 150°. Pyrrole-2-carboxyaldehyde (5% chemical yield) appeared first after the solvent. The single photo-product (55% chemical yield) isolated after the original 2-cyanopyrrole (retention time relative to 2-cyanopyrrole: 2.5) had the following spectral data; M 92; i.r. v_{max} 3430, 3280, 2230, 1250, 1090, and 1060 cm⁻¹ in CCl₄; n.m.r. three multiplets at τ 2.65 (1 H), 3.16 (1 H) 3.45 (1 H), and one broad singlet at τ 1.0 (1 H) in CCl₄.

2-Cyanopyrrole does not isomerize to the 3-isomer on heating for 4 hr at 180° in a methanol solution in a sealed tube. Therefore, the 2-3 isomerization is specific to photochemical reactions and different from the 1-2 and 1-3 valence isomerizations of the pyrroles.6,7

(Received, July 30th, 1970; Com. 1263.)

† The photo-dehydration has also been observed in furan-2-aldoxime; details will be published later.

- ² H. Hiraoka and R. Srinivasan, J. Amer. Chem. Soc., 1968, 90, 2720.
 ³ H. Wynberg, R. M. Kellogg, H. van Driel, and G. E. Beekhuis, J. Amer. Chem. Soc., 1967, 89, 3501, and earlier papers.
 ⁴ H. Tiefenthaler, W. Dörscheln, H. Göth, and H. Schmid, Helv. Chim. Acta, 1967, 50, 2444.

- ⁶ H. Hiraoka, J. Phys. Chem., 1970, 74, 574.
 ⁶ J. M. Patrerson, L. T. Burka, and M. R. Boyd, J. Org. Chem., 1968, 33, 4033.
 ⁷ J. M. Patterson and L. T. Burka, Tetrahedron Letters, 1969, 2215.
 ⁸ H. J. Anderson, Canad. J. Chem., 1959, 37, 2053.
 ⁹ H. J. Anderson, R. Marting, H. G. van der Plas, and J. Bon. Tetrahedron Letters, 1969, 2015.

- ⁹ H. J. den Hertog, R. J. Martens, H. C. van der Plas, and J. Bon, Tetrahedron Letters, 1966, 4325.

¹ E. F. Ullman and B. Singh, J. Amer. Chem. Soc., 1966, 88, 1844.